

## **Supporting information**

### **Potentiometric titration based on the reference electrode equipped with ionic liquid salt bridge**

#### **1. Precipitation titration of chloride with silver ions in water**

Takashi KAKIUCHI,<sup>a,b,\*,\$</sup> Ryunosuke TANIGO,<sup>a</sup> Atsushi TANI,<sup>a</sup> Takeshi YAMAZAKI,<sup>a</sup> Kohta KOMATSUBARA,<sup>a</sup> Keiji NAKANO,<sup>a</sup> and Masahiro YAMAMOTO<sup>a,\*,\$\$</sup>

*a Department of Chemistry, Konan University, 8-9-1 Okamoto, Higashinada-ku, Kobe, 658-8501, Japan*

*b pH Science and Technology Laboratory, Kinomoto 1058, Wakayama 640-8453, Japan*

*Corresponding authors: kakiuchi.takashi.55e@st.kyoto-u.ac.jp, (T.K.), masahiro@konan-u.ac.jp, (M.Y.)*

§ ECSJ Senior Member

\$\$ ECSJ Active Member

#### **ORCID**

Takashi Kakiuchi: 0000-0001-8603-4794

Masahiro Yamamoto: 0000-0002-3345-3917

Potentiometric titration of  $0.01 \text{ mol dm}^{-3}$  NaCl with  $0.01 \text{ mol dm}^{-3}$  AgNO<sub>3</sub> recorded with Cell (I) (dots) and least-square fitting curves of five models ( $\gamma_i \equiv 1$ (black)), DHLL(green), Bates-Guggenheim(purple), Güntelberg(blue), and Davies(red).

