Supporting Information

Development of a Molecular Recognition Electrode and Investigation of a Biomolecular Application in Non-Aqueous Media -Electrochemical Detection of Uremia-Related Substances Excreted via ATP-Binding Cassette Transporter G2-

Kyoko FUJITA,^{a,*, §§} Misaki SEKIDO,^a Kohei KANNO,^a Kio HATAE,^a and Kimiyoshi ICHIDA^a

^a Department of Pathophysiology, Tokyo University of Pharmacy and Life Sciences, 1432-1 Horinouchi, Hachioji, Tokyo, 192-0392 Japan

**Corresponding Author: <u>kyokof@toyaku.ac.jp</u>*

§§ ECSJ Active Member

ORCID ID:

Kyoko FUJITA: 0000-0002-0871-7896

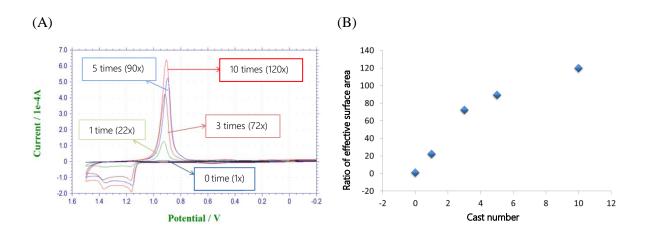


Figure S1 (A) Cyclic voltammograms of AuNP electrodes with different NP cast number in $0.1M H_2SO_4$ solution. 0 time means polycrystalline gold electrode (diameter: 5 mm) without AuNP. (B) Change in ratio of effective surface area related to the cast number.

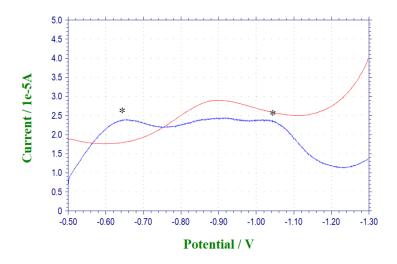


Figure S2 Cathodic stripping voltammetry in 0.5M KOH solution in the potential range -0.5 to -1.3 V. (Red: AuNP electrode without SAM modification, Blue: SAM-modified AuNP electrode)

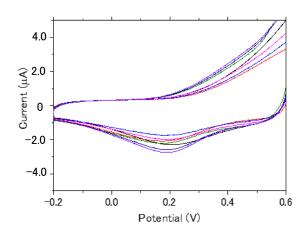


Figure S3 Cyclic voltammograms depending on IS addition $(0-6\mu g/mL)$ by 1 $\mu g/mL$ in phosphate buffer measured by non-pretreated 2-mercaptobenzimidazole modified AuNP electrode.

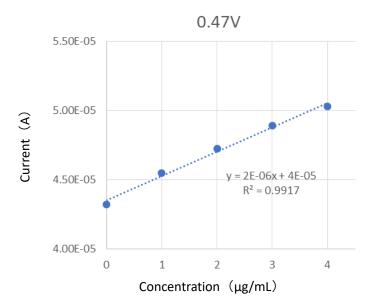


Figure S4 Standard curve of IS concentration vs oxidation current measured in serum-free culture medium.