

Supporting Information

Solid–Liquid Interfacial Properties Related to Ionic Conductivity of Mixtures of Metal Oxide Particles and Lithium Bis(fluorosulfonyl)amide-Sulfone Electrolytes

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Table S1. Median particle diameter (D50) and relative surface area of α -Al₂O₃ and LLZ powder used in this study.

| Material | D50 / μm | Relative surface area / $\text{m}^2 \text{g}^{-1}$ |
|--|---------------------|--|
| α -Al ₂ O ₃ | 27.8 | 0.52 |
| α -Al ₂ O ₃ | 14.8 | 0.46 |
| α -Al ₂ O ₃ | 4.46 | 1.4 |
| α -Al ₂ O ₃ | 0.804 | 2.3 |
| α -Al ₂ O ₃ | 0.110 | 51 |
| LLZ | 18.6 | 0.39 |
| LLZ | 9.31 | 0.71 |
| LLZ | 5.05 | 1.2 |
| LLZ | 0.872 | 4.0 |

Table S2. NMR measurement conditions of T_1 and T_2 .^a

Table S2. NMR measurement conditions of T_1 and T_2 .

| Samples | Nuclei | pulse sequence | x_pulse / μ s | Tau interval | | | Relaxation delay / s |
|--|-------------------------|--------------------|----------------------|--|-----|--------|-------------------------|
| | | | | T_1 : exp / s, T_2 : liner / ms | | | |
| | | | | Start | End | Points | |
| LiFSA:SL=1:3 | ^1H , T_1 | inversion recovery | 17.5 | 0.01 | 5 | 8 | 5 |
| | ^1H , T_2 | CPMG | 17.5 | 30 | 300 | 10 | 7 |
| | ^7Li , T_1 | inversion recovery | 9.4 | 0.01 | 5 | 8 | 5 |
| | ^7Li , T_2 | CPMG | 9.4 | 50 | 500 | 10 | 7 |
| | ^{19}F , T_1 | inversion recovery | 18.75 | 0.01 | 5 | 8 | 5 |
| | ^{19}F , T_2 | CPMG | 18.75 | 40 | 220 | 10 | 2 |
| α -Al ₂ O ₃ mixture ^b | ^1H , T_1 | inversion recovery | 15.4 | 0.01 | 10 | 8 | 10 |
| | ^1H , T_2 | CPMG | 15.4 | 0.40 | 4 | 10 | 8 |
| | ^7Li , T_1 | inversion recovery | 7.8 | 0.01 | 10 | 8 | 10 |
| | ^7Li , T_2 | CPMG | 7.8 | 0.40 | 4 | 10 | 10 |
| | ^{19}F , T_1 | inversion recovery | 17.3 | 0.10 | 5 | 8 | 5 |
| | ^{19}F , T_2 | CPMG | 17.3 | 0.40 | 4 | 10 | 1 |

^a Common conditions; 8 times scan, temperature of samples were 25 °C.^b (D50 0.11 μ m α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 61:39, t_{liquid} = 3.1 nm.

Table S3. NMR measurement conditions of PFG stimulated echo method.

| Samples | Diffusion time (Δ) / ms | PFG width (δ) / ms | PFG strength (G) ^a / T m ⁻¹ | | Gradient recovery times (τ) / ms | Echo time (T_e) / ms | Pulse width / μ s | Scans | Repetition time / s |
|--|--|-----------------------------------|--|------|--|-----------------------------------|-----------------------------|-------|------------------------|
| | | | start | end | | | | | |
| LiFSA: SL=1:3 | 500 | 1.0 | 0.1 | 5.0 | 0.5 | 50 | 8.5 | 8 | 2.4 |
| α -Al ₂ O ₃ mixture ^b | 12 | 1.2 | 0.1 | 13.5 | 0.2 | 20 | 8.5 | 1024 | 2.0 |

^a log array (base number: 2), 8 points

^b (D50 0.11 μ m α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 61:39, t_{liquid} = 3.1 nm.

Table S4. NMR measurement conditions of $^7\text{Li} \{^1\text{H}\}$ HOESY and $^7\text{Li} \{^1\text{H}\}$ LGCP-HETCOR.

| Samples | Method | Conditions | | | | | |
|---|---|-----------------------------------|--------------------------|--------|--------------|-------|------------|
| LiFSA: SL=1:3 | $^7\text{Li} \{^1\text{H}\}$ HOESY | x | y | y | Mixing | Scans | Repetition |
| | | Pulse / μs | Pulse / μs | Points | time / s | | |
| | | 10 | 15 | 128 | 1 | 16 | 2 |
| $\alpha\text{-Al}_2\text{O}_3\text{-}$ (LiFSA: SL=1:3) ^a | $^7\text{Li} \{^1\text{H}\}$ LGCP- HETCOR | Irradiation | y | y | Contact | Scans | Repetition |
| | | pulse width / μs | Acquisition time / ms | Points | time / ms | | |
| | | 3.15 | 5.25 | 64 | 3 | 8 | 2 |

^a (D50 0.11 μm $\alpha\text{-Al}_2\text{O}_3$):(LiFSA:SL=1:3) = volume ratio 61:39, $t_{\text{liquid}} = 3.1$ nm.

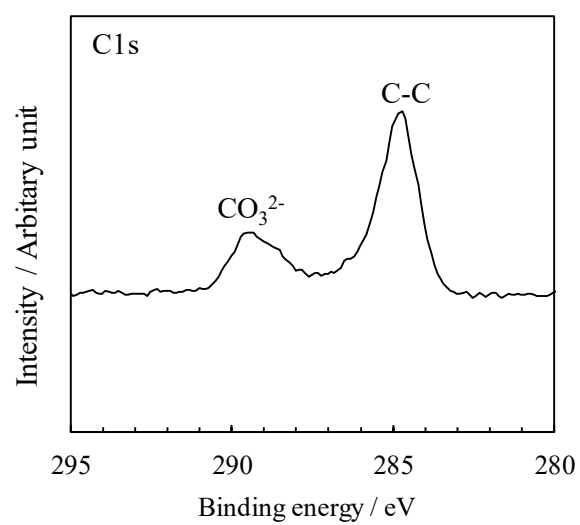


Figure S1. XPS (C1s) of LLZ particle surface.

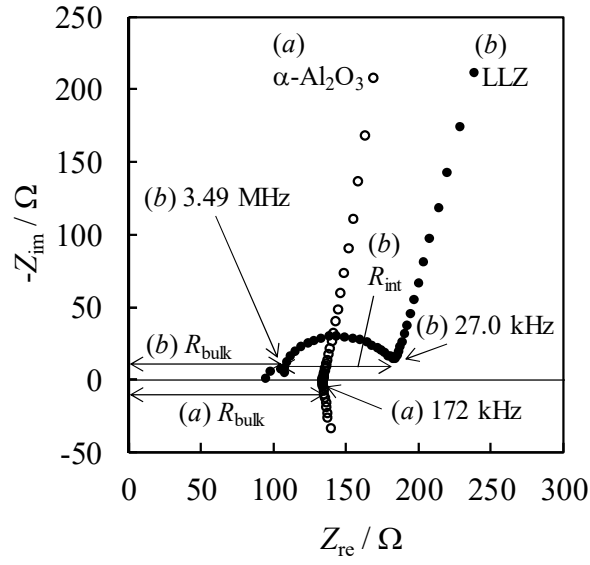


Figure S2. R_{bulk} and R_{int} evaluation of metal oxide mixture pellets (diameter: 10 mm).

(a) (D50 0.804 μm $\alpha\text{-Al}_2\text{O}_3$):(LiFSA:SL = 1:3) = volume ratio 69:31,

$t_{\text{liquid}} = 69$ nm, pellet thickness: 0.53 mm, and

(b) (D50 0.872 μm LLZ):(LiFSA:SL = 1:3) = volume ratio 60:40,

$t_{\text{liquid}} = 32$ nm, pellet thickness: 0.54 mm.

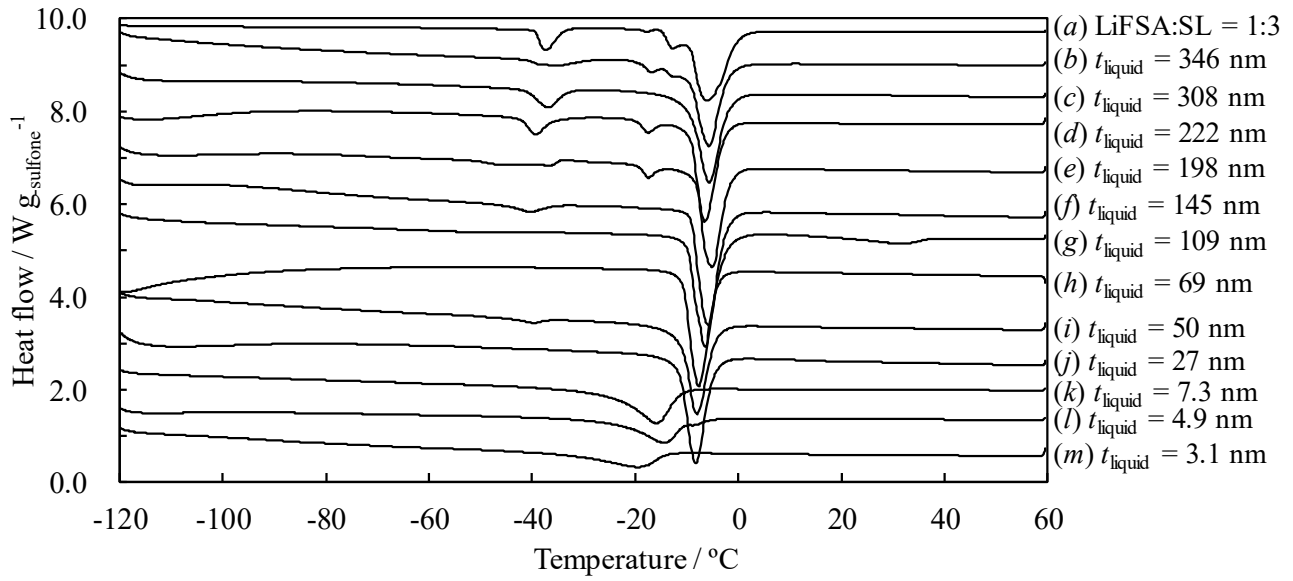


Figure S3. DSC curves of mixture of α -Al₂O₃ and (LiFSA:SL = 1:3) with different t_{liquid} samples.

(a) LiFSA:SL = 1:3 (without α -Al₂O₃),

(b) (D50 14.8 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 61:39, t_{liquid} = 346 nm,

(c) (D50 27.8 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 61:39, t_{liquid} = 308 nm,

(d) (D50 14.8 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 71:29, t_{liquid} = 222 nm,

(e) (D50 27.8 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 71:29, t_{liquid} = 198 nm,

(f) (D50 14.8 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 79:21, t_{liquid} = 145 nm,

(g) (D50 0.804 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 50:50, t_{liquid} = 109 nm,

(h) (D50 0.804 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 61:39, t_{liquid} = 69 nm,

(i) (D50 0.804 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 69:31, t_{liquid} = 50 nm,

(j) (D50 0.804 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 80:20, t_{liquid} = 27 nm,

(k) (D50 0.110 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 40:60, t_{liquid} = 7.3 nm,

(l) (D50 0.110 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 50:50, t_{liquid} = 4.9 nm,

and (m) (D50 0.110 μm α -Al₂O₃):(LiFSA:SL = 1:3) = volume ratio 61:39, t_{liquid} = 3.1 nm.

Heat flow is based on SL weight (g-sulfone), temperature scan rate: 10 °C min⁻¹.

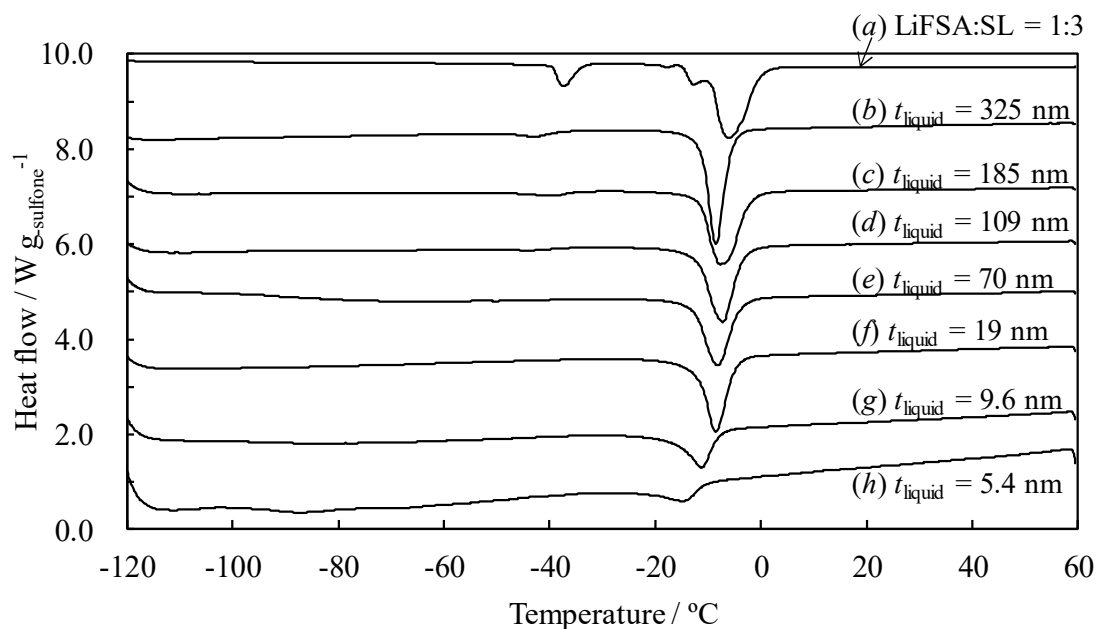


Figure S4. DSC curves of mixture of LLZ and (LiFSA:SL = 1:3) with different t_{liquid} samples.

(a) LiFSA:SL=1:3 (without LLZ),

(b) (D50 18.6 μm LLZ):(LiFSA:SL = 1:3) = volume ratio 61:39, $t_{\text{liquid}} = 325$ nm,

(c) (D50 9.31 μm LLZ):(LiFSA:SL = 1:3) = volume ratio 60:40, $t_{\text{liquid}} = 185$ nm,

(d) (D50 5.05 μm LLZ):(LiFSA:SL = 1:3) = volume ratio 60:40, $t_{\text{liquid}} = 109$ nm,

(e) (D50 5.05 μm LLZ):(LiFSA:SL = 1:3) = volume ratio 70:30, $t_{\text{liquid}} = 70$ nm,

(f) (D50 0.872 μm LLZ):(LiFSA:SL = 1:3) = volume ratio 72:28, $t_{\text{liquid}} = 19$ nm,

(g) (D50 0.872 μm LLZ):(LiFSA:SL = 1:3) = volume ratio 84:16, $t_{\text{liquid}} = 9.6$ nm,

and (h) (D50 0.872 μm LLZ):(LiFSA:SL = 1:3) = volume ratio 90:10, $t_{\text{liquid}} = 5.4$ nm.

Heat flow is based on SL weight ($\text{g}_{\text{-sulfone}}$), temperature scan rate: $10\text{ }^{\circ}\text{C min}^{-1}$.

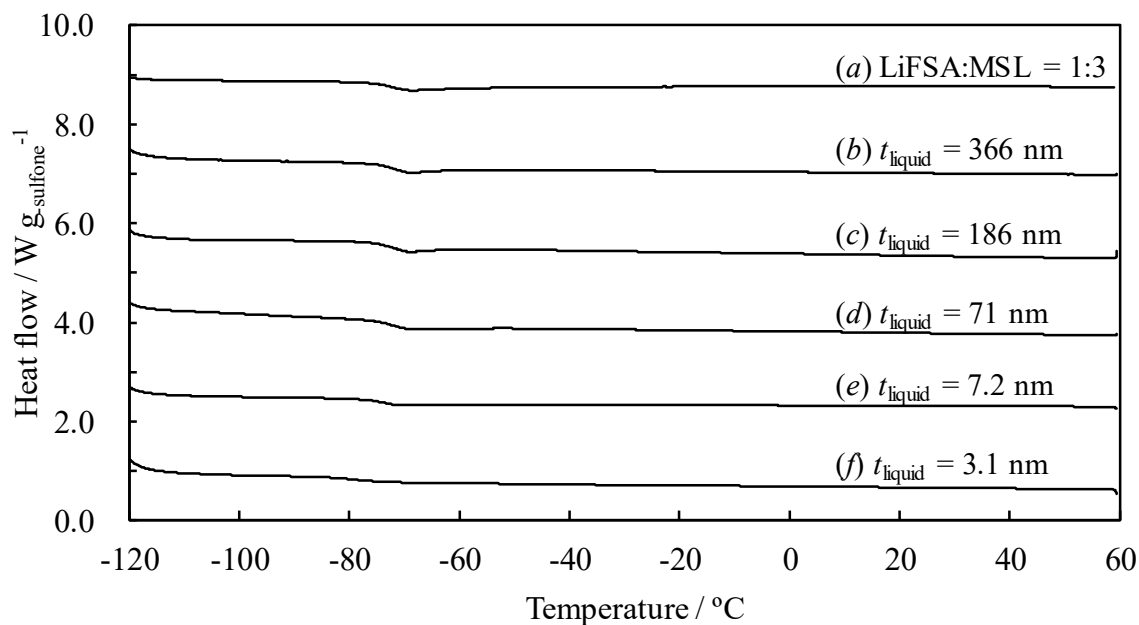


Figure S5. DSC curves of mixture of α -Al₂O₃ and (LiFSA:MSL = 1:3) with different t_{liquid} samples.

(a) LiFSA:MSL = 1:3 (without α -Al₂O₃),

(b) (D50 14.8 μm α -Al₂O₃):(LiFSA:MSL = 1:3) = volume ratio 60:40, t_{liquid} = 366 nm,

(c) (D50 14.8 μm α -Al₂O₃):(LiFSA:MSL = 1:3) = volume ratio 75:25, t_{liquid} = 186 nm,

(d) (D50 0.804 μm α -Al₂O₃):(LiFSA:MSL = 1:3) = volume ratio 61:39, t_{liquid} = 71 nm,

(e) (D50 0.110 μm α -Al₂O₃):(LiFSA:MSL = 1:3) = volume ratio 41:59, t_{liquid} = 7.2 nm,

and (f) (D50 0.110 μm α -Al₂O₃):(LiFSA:MSL = 1:3) = volume ratio 61:39, t_{liquid} = 3.1 nm.

Heat flow is based on MSL weight ($\text{g}_{\text{sulfone}}$), temperature scan rate: 10 $^{\circ}\text{C min}^{-1}$.

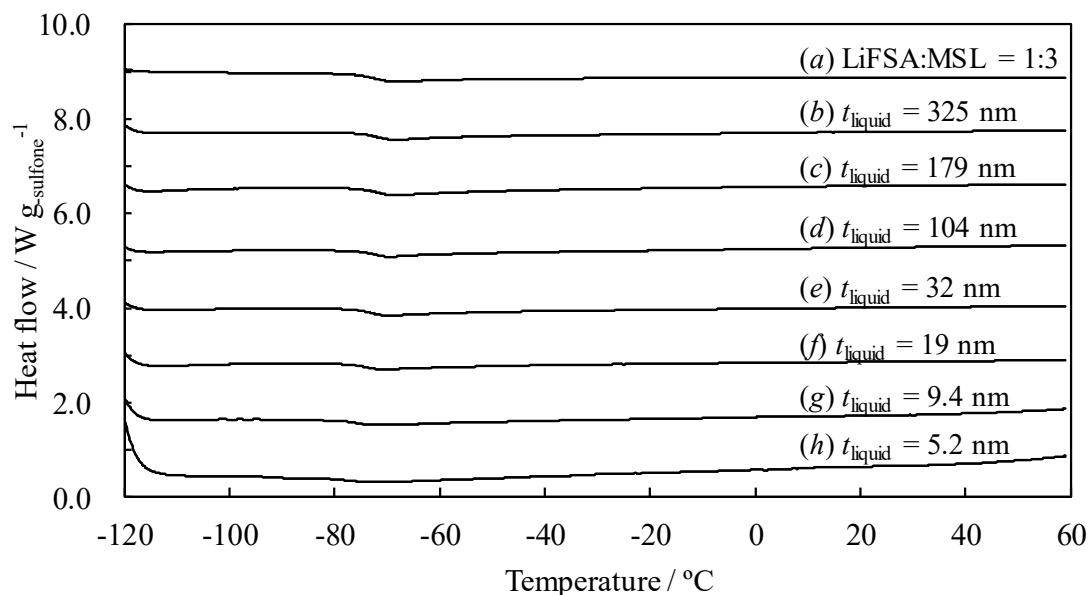


Figure S6. DSC curves of mixture of LLZ and (LiFSA:MSL = 1:3) with different t_{liquid} samples.

(a) LiFSA:MSL = 1:3 (without LLZ),

(b) (D50 18.6 μm LLZ):(LiFSA:MSL = 1:3) = volume ratio 61:39, t_{liquid} = 325 nm,

(c) (D50 9.31 μm LLZ):(LiFSA:MSL = 1:3) = volume ratio 61:39, t_{liquid} = 179 nm,

(d) (D50 5.05 μm LLZ):(LiFSA:MSL = 1:3) = volume ratio 61:39, t_{liquid} = 104 nm,

(e) (D50 0.872 μm LLZ):(LiFSA:MSL = 1:3) = volume ratio 61:39, t_{liquid} = 32 nm,

(f) (D50 0.872 μm LLZ):(LiFSA:MSL = 1:3) = volume ratio 72:28, t_{liquid} = 19 nm,

(g) (D50 0.872 μm LLZ):(LiFSA:MSL = 1:3) = volume ratio 84:16, t_{liquid} = 9.4 nm,

and (h) (D50 0.872 μm LLZ):(LiFSA:MSL = 1:3) = volume ratio 90:10, t_{liquid} = 5.2 nm.

Heat flow is based on MSL weight ($\text{g}_{\text{sulfone}}$), temperature scan rate: $10\text{ }^{\circ}\text{C min}^{-1}$.

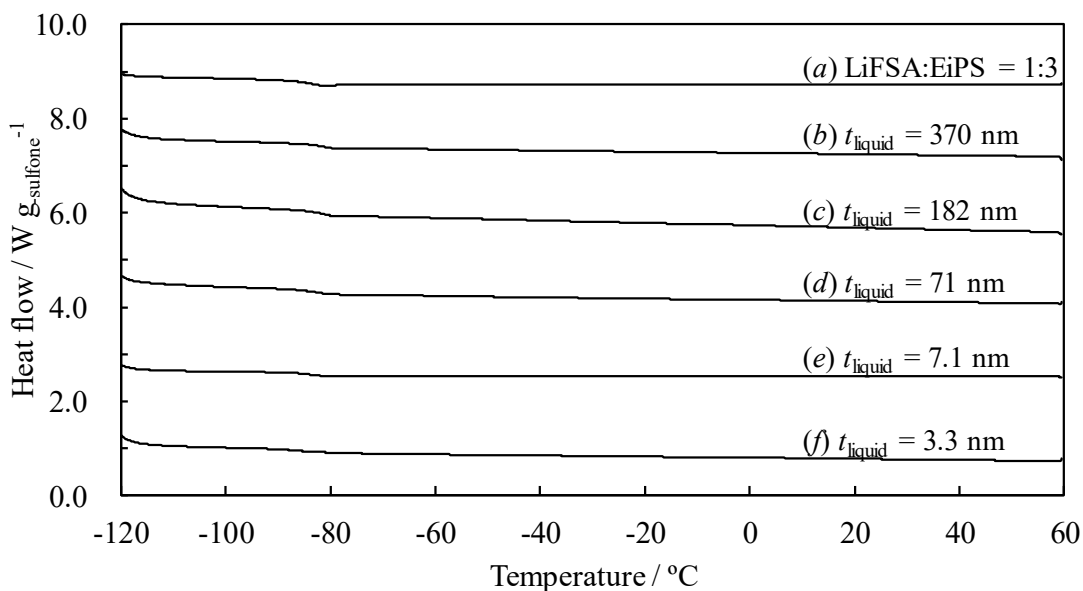


Figure S7. DSC curves of mixture of α -Al₂O₃ and (LiFSA:EiPS = 1:3) with different t_{liquid} samples.

(a) LiFSA:EiPS = 1:3 (without α -Al₂O₃),

(b) (D50 14.8 μm α -Al₂O₃):(LiFSA:EiPS = 1:3) = volume ratio 60:40, t_{liquid} = 370 nm,

(c) (D50 14.8 μm α -Al₂O₃):(LiFSA:EiPS = 1:3) = volume ratio 75:25, t_{liquid} = 182 nm,

(d) (D50 0.804 μm α -Al₂O₃):(LiFSA:EiPS = 1:3) = volume ratio 60:40, t_{liquid} = 71 nm,

(e) (D50 0.110 μm α -Al₂O₃):(LiFSA:EiPS = 1:3) = volume ratio 41:59, t_{liquid} = 7.1 nm,

and (f) (D50 0.110 μm α -Al₂O₃):(LiFSA:EiPS = 1:3) = volume ratio 60:40, t_{liquid} = 3.3 nm.

Heat flow is based on EiPS weight (g-sulfone), temperature scan rate: 10 $^{\circ}\text{C min}^{-1}$.

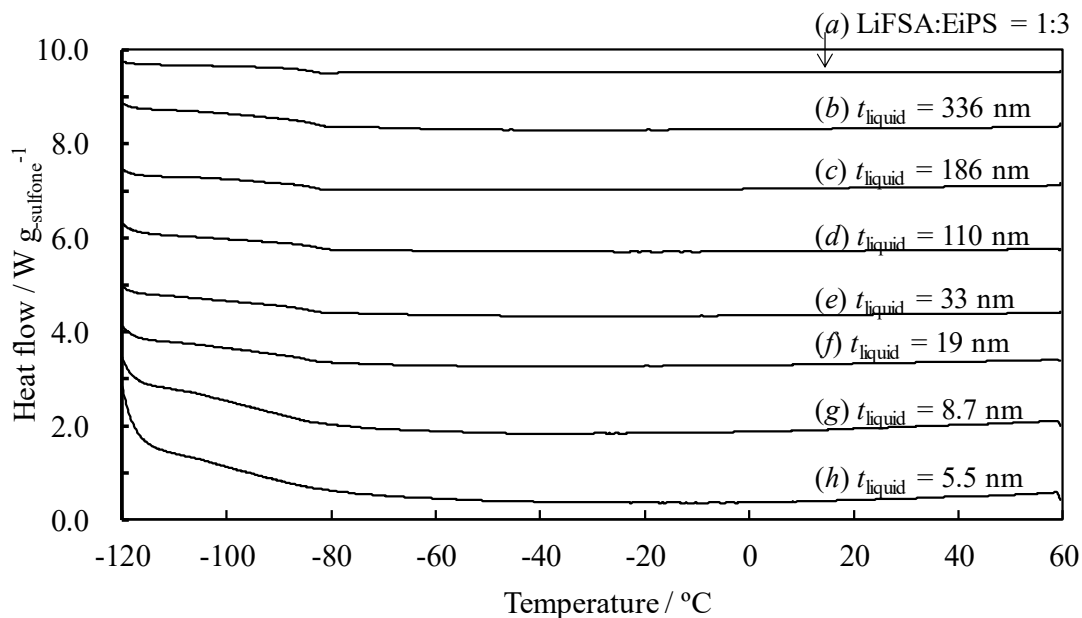


Figure S8. DSC curves of mixture of LLZ and (LiFSA:EtPS = 1:3) with different t_{liquid} samples.

(a) LiFSA:EtPS = 1:3 (without LLZ),

(b) (D50 18.6 μm LLZ):(LiFSA:EtPS = 1:3) = volume ratio 60:40, $t_{\text{liquid}} = 336 \text{ nm}$,

(c) (D50 9.31 μm LLZ):(LiFSA:EtPS = 1:3) = volume ratio 60:40, $t_{\text{liquid}} = 186 \text{ nm}$,

(d) (D50 5.05 μm LLZ):(LiFSA:EtPS = 1:3) = volume ratio 60:40, $t_{\text{liquid}} = 110 \text{ nm}$,

(e) (D50 0.872 μm LLZ):(LiFSA:EtPS = 1:3) = volume ratio 60:40, $t_{\text{liquid}} = 33 \text{ nm}$,

(f) (D50 0.872 μm LLZ):(LiFSA:EtPS = 1:3) = volume ratio 72:28, $t_{\text{liquid}} = 19 \text{ nm}$,

(g) (D50 0.872 μm LLZ):(LiFSA:EtPS = 1:3) = volume ratio 85:15, $t_{\text{liquid}} = 8.7 \text{ nm}$,

and (h) (D50 0.872 μm LLZ):(LiFSA:EtPS = 1:3) = volume ratio 90:10, $t_{\text{liquid}} = 5.5 \text{ nm}$.

Heat flow is based on EtPS weight (g_{sulfone}), temperature scan rate: 10 °C min⁻¹.

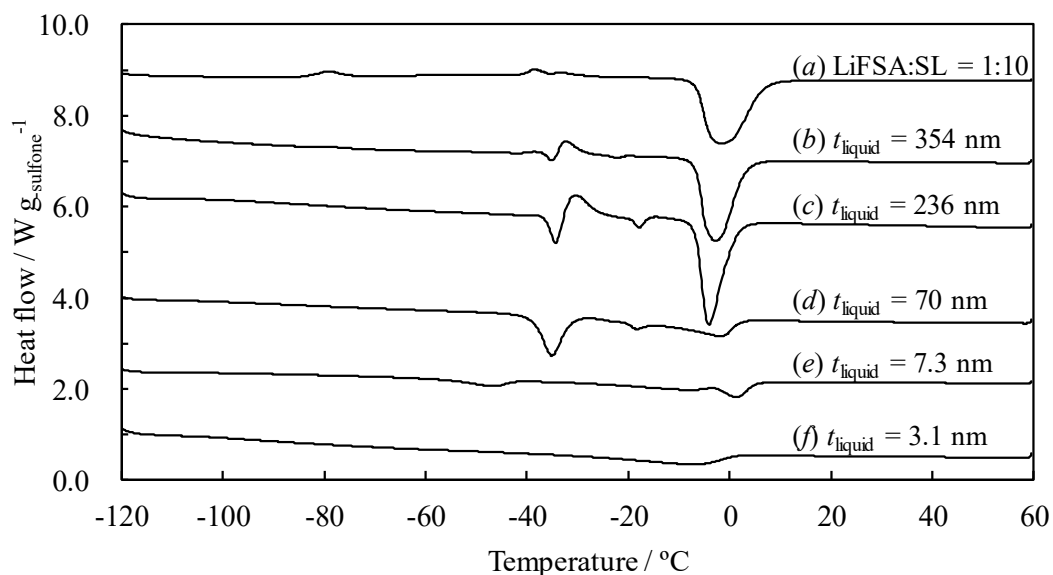


Figure S9. DSC curve of mixture of α -Al₂O₃ and (LiFSA:SL = 1:10) with different t_{liquid} samples.

(a) LiFSA:SL = 1:10 (without α -Al₂O₃),

(b) (D50 14.8 μm α -Al₂O₃):(LiFSA:SL = 1:10) = volume ratio 61:39, $t_{\text{liquid}} = 354$ nm,

(c) (D50 14.8 μm α -Al₂O₃):(LiFSA:SL = 1:10) = volume ratio 70:30, $t_{\text{liquid}} = 236$ nm,

(d) (D50 0.804 μm α -Al₂O₃):(LiFSA:SL = 1:10) = volume ratio 61:39, $t_{\text{liquid}} = 70$ nm,

(e) (D50 0.110 μm α -Al₂O₃):(LiFSA:SL = 1:10) = volume ratio 40:60, $t_{\text{liquid}} = 7.3$ nm,

and (f) (D50 0.110 μm α -Al₂O₃):(LiFSA:SL = 1:10) = volume ratio 61:39, $t_{\text{liquid}} = 3.1$ nm.

Heat flow is based on SL weight (g-sulfone), temperature scan rate: 10 °C min⁻¹.

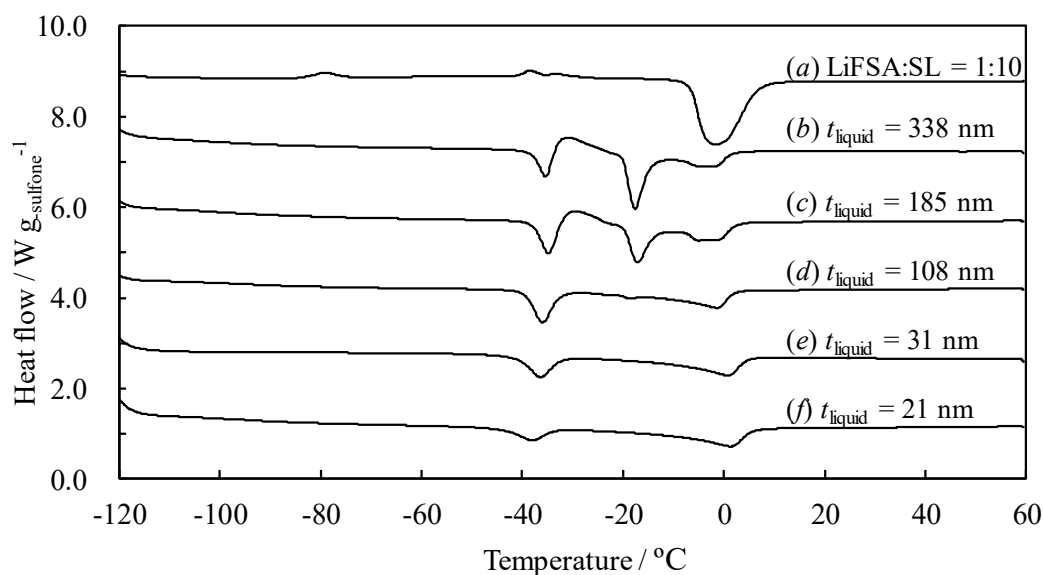


Figure S10. DSC curve of mixture of LLZ and (LiFSA:SL = 1:10) with different t_{liquid} samples.

(a) LiFSA:SL = 1:10 (without LLZ),

(b) (D50 18.6 μm LLZ):(LiFSA:SL = 1:10) = volume ratio 60:40, $t_{\text{liquid}} = 338 \text{ nm}$,

(c) (D50 9.31 μm LLZ):(LiFSA:SL = 1:10) = volume ratio 60:40, $t_{\text{liquid}} = 185 \text{ nm}$,

(d) (D50 5.05 μm LLZ):(LiFSA:SL = 1:10) = volume ratio 60:40, $t_{\text{liquid}} = 108 \text{ nm}$,

(e) (D50 0.872 μm LLZ):(LiFSA:SL = 1:10) = volume ratio 61:39, $t_{\text{liquid}} = 31 \text{ nm}$,

and (f) (D50 0.872 μm LLZ):(LiFSA:SL = 1:10) = volume ratio 70:30, $t_{\text{liquid}} = 21 \text{ nm}$.

Heat flow is based on SL weight ($\text{g}_{\text{sulfone}}$), temperature scan rate: $10^{\circ}\text{C min}^{-1}$.

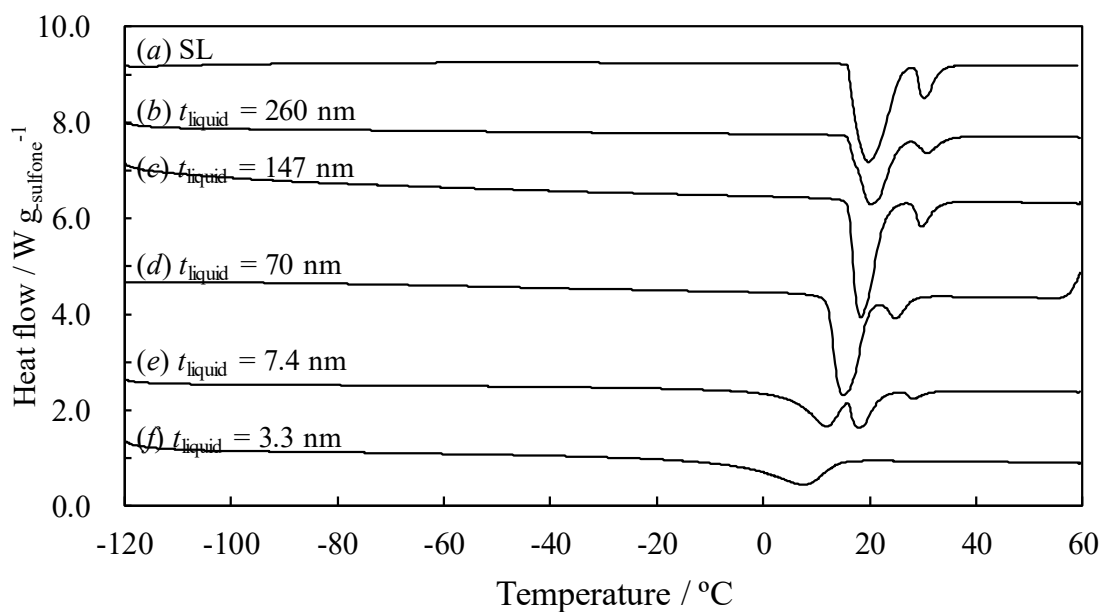


Figure S11. DSC curve of mixture of α -Al₂O₃ and SL with different t_{liquid} samples.

(a) SL (without α -Al₂O₃),

(b) (D50 14.8 μm α -Al₂O₃):SL = volume ratio 68:32, $t_{\text{liquid}} = 260$ nm,

(c) (D50 14.8 μm α -Al₂O₃):SL = volume ratio 79:21, $t_{\text{liquid}} = 147$ nm,

(d) (D50 0.804 μm α -Al₂O₃):SL = volume ratio 61:39, $t_{\text{liquid}} = 70$ nm,

(e) (D50 0.110 μm α -Al₂O₃):SL = volume ratio 40:60, $t_{\text{liquid}} = 7.4$ nm,

and (f) (D50 0.110 μm α -Al₂O₃):SL = volume ratio 60:40, $t_{\text{liquid}} = 3.3$ nm.

Heat flow is based on SL weight (g-sulfone), temperature scan rate: $10\text{ }^{\circ}\text{C min}^{-1}$.

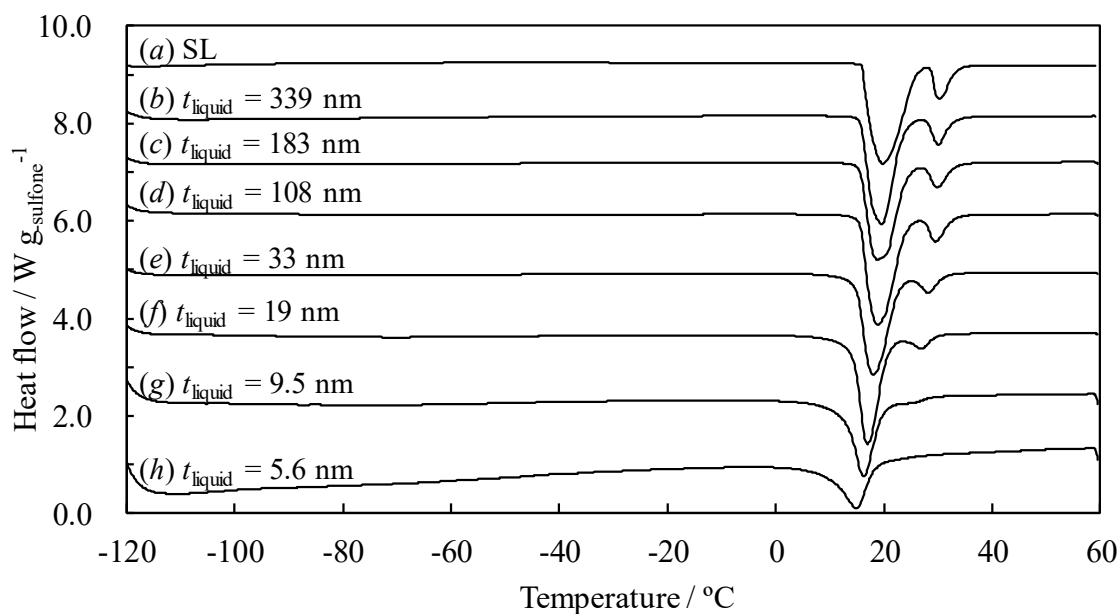


Figure S12. DSC curve of mixture of LLZ and SL with different t_{liquid} samples.

(a) SL (without LLZ),

(b) (D50 18.6 μm LLZ):SL = volume ratio 60:40, $t_{\text{liquid}} = 339$ nm,

(c) (D50 9.31 μm LLZ):SL = volume ratio 60:40, $t_{\text{liquid}} = 183$ nm,

(d) (D50 5.05 μm LLZ):SL = volume ratio 60:40, $t_{\text{liquid}} = 108$ nm,

(e) (D50 0.872 μm LLZ):SL = volume ratio 60:40, $t_{\text{liquid}} = 33$ nm,

(f) (D50 0.872 μm LLZ):SL = volume ratio 72:28, $t_{\text{liquid}} = 19$ nm,

(g) (D50 0.872 μm LLZ):SL = volume ratio 84:16, $t_{\text{liquid}} = 9.5$ nm,

and (h) (D50 0.872 μm LLZ):SL = volume ratio 90:10, $t_{\text{liquid}} = 5.6$ nm.

Heat flow is based on SL weight ($\text{g}_{\text{sulfone}}$), temperature scan rate: $10\text{ }^{\circ}\text{C min}^{-1}$.